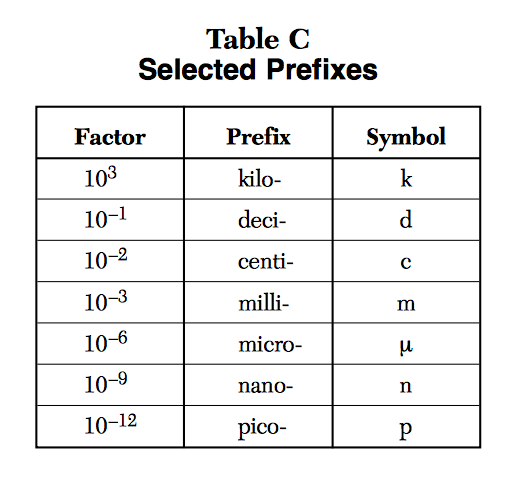
Name: \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ Per. \_\_\_\_\_\_\_\_

2018 Q1Q2 Metric B



1. Describe how you would move the decimal when converting μm to mm?

2. Describe how you would move the decimal when converting J to kJ?

3. Describe how you would move the decimal when converting m to cm?

4. Describe how you would move the decimal when converting pg to g?

5. What is the factor value for any base unit missing from this table?

**For questions 6 – 8**: Complete the metric conversion by unit analysis.

\*\* You can use the example below to help you show the proper setup and solve.

REMEMBER to Keep sig figs the same!

**Example**: 2.35 km = \_\_\_\_\_\_\_\_\_\_\_\_ m the table tells us *1 km = 1000 m*

2.35 km x 1000 m = 2350 m (**km cancels out** and notice decimal

1 km moved 3 places to the right)

6. .0550 m = \_\_\_\_\_\_\_\_\_\_\_\_ cm

7. .000350 kJ = \_\_\_\_\_\_\_\_\_\_\_\_ J

8. How many μL are in 2500 mL?

Questions 9 – 22: Complete the DRUL metric conversion.

\*\* **REMEMBER to Keep sig figs the same**!

9. Which of the following is the *longest* length?

1. 2.710 x 103 m 3. 27100000 μm

2. 27.10 km 4. 27100 cm

10. 1013000 Pa = \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ kPa

11. 3500 J = \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ kJ

12. 750 mL = \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ dL

13. 500. g = \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ kg

14. .09500 g = \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ μg

15. .00025 mmol = \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ μmol

16. 920000 ng = \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ pg

17. **2.09 x 10-4**m = \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ mm

18. **6.25 x 107** μL = \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ L

19. .0000325 J = \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_μJ

20. 3.270 x 1012 pm = \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ km

\*\*HONORS OPTION\*\*

21. If 1 mL = 1 cm3 = 1 cc then 20000 cc = \_\_\_\_\_\_\_\_\_\_\_\_\_μL

22. If 1 Cal = 1000 cal then 1120 Cal = \_\_\_\_\_\_\_\_\_\_\_\_\_\_ cal